# **NEOLIGNANS FROM MEZILAURUS ITAUBA\***

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Abstract—The wood bark of *Mezilaurus itauba* afforded in addition to seven known neolignans, three new compounds rel-  $(7R,8R,1'S,3'S)-\Delta^{5',8'}-5'$ -methoxy-3,4-methylenedioxy-1',2',3',4'-tetrahydro-2',4'-dioxo-7.3',8.1'-neolignan, rel-  $(7S,8S,1'S,2'S,3'R,4'S)-\Delta^{8'}-2',4'$ -dihydroxy-3,4-methylenedioxy-1',2',3',4',5',6'-hexahydro-5'-oxo-7.3',8.1'-neolignan and rel-  $(7S,8S)-\Delta^{8'}-6'$ -hydroxy-5'-methoxy-3,4-methylenedioxy-7·O-2',8.3'-neolignan. The latter compound has been detected previously in *Aniba terminalis*. The structures were elucidated by spectroscopic methods and comparison with related compounds.

#### INTRODUCTION

Mezilaurus is a genus of the Lauraceae; comprising nine mostly arboreal species distributed mainly in the Amazon basin [1]. Only M. synandra has been investigated chemically and its trunk wood contains benzylisoquinoline alkaloids and  $\gamma$ -lactones [2].

The present paper describes the results obtained in a study of *M. itauba* (Meissn.) Taubert ex Mez., which grows in the Comisarías del Amazonas and Guaviare, Colombia; the plant is commonly known as itauba-stone wood, 'Japi-u-dlr' and 'Macmemorae' (dialects: Puinave and Witoto, respectively) [3]. The chloroform extract of the wood bark of *M. itauba* yielded seven benzofuranoid neolignans and three bicyclo[3.2.1]octanoid neolignans. The nomenclature and numbering of these neolignans follow the rules which were outlined in a recent review [4].

## RESULTS AND DISCUSSION

Ten neolignans were isolated from *M. itauba*. Among these compounds 1b, 2a, 4a and 5a were reported earlier from *Aniba terminalis* [5, 6], neolignans of structural types 1b, 3a and 6a co-occur in *A. burchellii* [7, 8] and 7a was previously isolated from an *Aniba* species [9].

Compound 1a showed IR absorption at  $3500\,\mathrm{cm}^{-1}$  indicative of a hydroxyl group; confirmation of the presence of this functionality was provided by the UV spectrum ( $\lambda_{\max}^{MeOH}$  325 nm;  $\lambda_{\max}^{MeOH+NeOMe}$  344 nm). The complex UV spectrum of 1a, is indicative of a highly aromatic structure. Particularly informative was the <sup>1</sup>H NMR spectrum measured in carbon tetrachloride which showed a hydroxyl signal as a broad peak at  $\delta$ 5.51, three aromatic protons as a multiplet at  $\delta$ 6.68-6.90 and one aromatic proton singlet at  $\delta$ 6.42. The AMX<sub>3</sub> system was analysed by <sup>1</sup>H NMR; (A):  $\delta$ 4.93 (d, J = 8.0 Hz); (M):

For compound 7b,  $C_{20}H_{20}O_5$  ([M]<sup>+</sup> m/z 340),  $[\alpha]_D$ -7.31°, the formula, established by mass spectrometry and expanded by inspection of UV, IR and <sup>1</sup>H NMR spectra, showed that the two C<sub>6</sub>-C<sub>3</sub> units in this neolignan are linked only by C-C bonds and suggest that the compound is a bicyclo-octanoid [4]. Mass spectra were very helpful in defining one of the C<sub>6</sub>-C<sub>3</sub> units by the fragment at m/z 162 (47%) which corresponded to an [ArCH=CHMe] t fragment. The constitution of the second C<sub>6</sub>-C<sub>3</sub> unit was defined by spectroscopic methods. Compound 7b showed IR absorptions at 1770 and 1718 cm<sup>-1</sup> (>CO) corresponding to part of the fivemembered and six-membered rings, respectively. The stereochemical assignments of 7b are corroborated by <sup>1</sup>H NMR spectral comparisons with a model compound [10]

Compound 8a is a crystalline compound, mp 158–160°,  $[\alpha]_D - 0.69^\circ$ ,  $[M]^+ m/z$  330 showing aromatic ring absorption at  $\lambda_{max}$  289 nm (log  $\epsilon$ 3.51) in the UV spectrum and carbonyl absorption at 1698 cm<sup>-1</sup> in the IR spectrum. Absorption at 3504–3378 cm<sup>-1</sup> is attributed to hydroxyl groups and bands at 1640, 985 and 921 cm<sup>-1</sup> indicate the presence of an allyl group. The aliphatic nature of the hydroxyl group was deduced by inspection of the <sup>13</sup>C and <sup>1</sup>H NMR spectra, which in addition gave further structural information (Table 1). C-2' and C-4' are represented by doublets at  $\delta$ 79.0 and  $\delta$ 75.7 and the two carbinol protons resonate at  $\delta$ 4.60–4.76 (m, H-2') and

 $<sup>\</sup>delta 3.10-3.50$  (m); (X<sub>3</sub>):  $\delta 1.35$  (d, J=7.0 Hz). A trans-7,8-dihydrobenzofuran structure of 1a, was suggested by correlation with representatives of established structure [7]. An allyl group is also evident from the <sup>1</sup>H NMR spectrum which also exhibited signals due to methylene ( $\delta 3.38$ , d, J=6.0 Hz) and vinyl ( $\delta 5.60-6.20$ , m, =CH and  $\delta 4.80-5.20$ , m, =CH<sub>2</sub>) groups as well as signals due to methoxy ( $\delta 3.83$ , s) and methylenedioxy ( $\delta 5.90$ , s) groups. Additional structural evidence was gained by preparation of 1c. Convincing evidence for the structure of 1a was provided by the mass spectrum ([M]<sup>+</sup> at m/z 340 and fragment at m/z 162). This compound has been previously detected in a mixture [5].

<sup>\*</sup>Based in part on the M.Sc. thesis submitted by X. Y. R. to Universidad Nacional de Colombia, Bogotá, Colombia (1985). †To whom correspondence should be addressed.

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3.76–3.96 (m, H-4'). The aromatic ring is evident from two  $^{13}$ C oxygen bearing singlets at  $\delta$ 148.7 and 147.6, three doublets at  $\delta$ 108.1, 109.5 and 122.3 and a singlet at  $\delta$ 140.4. This is confirmed by signals for three aromatic protons at  $\delta$ 7.3 (d, J=2.0 Hz, H-2), 6.78 (d, J=6.0 Hz, H-5) and 6.78 (dd, J=6.0 and 2.0 Hz) in the  $^1$ H NMR spectrum which shows a signal for one methylenedioxy group at  $\delta$ 5.96 (s). The remaining  $sp^2$   $^{13}$ C signals are ascribed to a carbonyl group at  $\delta$ 210.5 and a monosubsti-

tuted double bond responsible for a doublet at  $\delta 136.3$  (C-8') and a triplet at 117.8 (C-9'). The substituent of the double bond is a methylene group responsible for a multiplet ( $\delta 1.7-2.2$ ), coupled to a vinyl proton that appears as a multiplet ( $\delta 5.6-6.2$ , H-8') further coupled to two vinyl protons ( $\delta 4.88-5.36$ , m, 2H-9'). In the <sup>13</sup>C NMR spectrum C-7' appeared as a triplet at  $\delta 38.6$ . The doublet at  $\delta 0.87$  (J = 7.0 Hz, 3H-9) indicate an Ar-7/Me-8-trans-relationship where the methyl is endo with

<u>8b</u>

Table 1. 13C and 1H chemical shifts for 8a

	<sup>13</sup> C		1
Assignment	δMe <sub>2</sub> CO-d <sub>6</sub> *	APT†	<sup>1</sup> H δMe <sub>2</sub> CO-d <sub>6</sub> ‡
1	140.4	С	_
2	108.1	CH	7.30
3	148.7	С	_
4	147.6	С	_
5	109.5	CH	6.78
6	122.3	CH	6.78
7	55.2	CH	2.70
8	49.2	CH	2.20-2.60
9	12.3	Me	0.87
1'	51.7	С	_
2'	79.0	CH	4.60-4.76
3'	55.9	CH	2.20-2.60
4'	75.7	CH	3.76-3.96
5'	210.5	С	_
6'	43.4	CH <sub>2</sub>	2.20-2.60
7'	38.6	CH,	1.70-2.20
8'	136.3	CH	5.60-6.20
9′	117.8	CH <sub>2</sub>	4.88-5.36
O <sub>2</sub> CH <sub>2</sub>	101.6	CH <sub>2</sub>	5.96

- \*Measured at 20.0 MHz with TMS as internal reference.
- †Information obtained from the APT.
- ‡Measured at 60 MHz with TMS as internal reference.

respect to bicyclo [11]. The clearest evidence for the *endo* orientation of C-9 and C-6' was given by  $^{13}$ C NMR signals at  $\delta$ 12.3 (C-9) and  $\delta$ 43.4 (C-6') corresponding to height values by the reciprocal  $\gamma$ -effect [12].

Acetylation transformed 8a into the triacetate 8b. A low resolution mass spectrum obtained at 30 eV, with significant peaks at m/z 456 (100%) [M]<sup>+</sup>, 414 (5%), 355 (5%), 296 (3%) and 162 (59%), confirmed this reaction. This information, the strong IR band at 1748 cm<sup>-1</sup>, the disappearance of the absorption at 1698 cm<sup>-1</sup> present in 8a and the number and chemical shifts of <sup>13</sup>C and <sup>1</sup>H NMR signals suggested the formation of acetate 8b with enolization of 8a. The acetate 8b contains three acetoxy groups localized in 2' ( $\delta$ 2.08, s, 3H;  $\delta$ 20.86, q, 1C and  $\delta$ 169.38, s, 1C), 4' ( $\delta$ 2.13, s, 3H;  $\delta$ 21.04, q, 1C and  $\delta$ 169.79, s, 1C) and 5' ( $\delta$ 2.16, s, 3H;  $\delta$ 21.29, q, 1C and  $\delta$ 170.39, s, 1C). Further inspection of the <sup>1</sup>H and <sup>13</sup>C NMR spectra of **8a** and **8b** is of particular relevance. In 8b there are four olefinic protons (8a contains three olefinic protons) and the multiplet at  $\delta$ 2.2-2.6 present in 8a is absent. In the <sup>13</sup>C NMR spectrum there is disappearance of the signals at  $\delta 43.4$  (8a: t. 1C) and  $\delta 210.50$  (8a: s, 1C) and the appearance of a doublet at  $\delta$  121.44 (C-6') and a singlet at  $\delta$  147.93 (C-5'). Preliminary information about the number of directly attached protons to each carbon atom was obtained from APT (attached proton test) experiments (Table 1). Spectral comparison of 8a and of its triacetate 8b with canellin-C [11, 13] and other related compounds [9] indicates that 8a and canellin-C to possess identical configurations.

## **EXPERIMENTAL**

The plant material from the Bajo Guaviare, Comisaría del Guainía (Colombia), voucher Xiloteca de la Facultad de

Ingeniería Forestal, Universidad Distrital, Bogotá, Colombia, 424, was collected and identified by Drs. A. Roa and E. Acero.

Isolation of constituents. Wood bark (2 kg) was extracted with EtOH. The CHCl<sub>3</sub> soluble part (20 g) of the extract (76 g) was chromatographed on a silica S column (200 g). Fractions eluted with  $C_6H_6$  were rechromatographed and further purified by prep. TLC (petrol- $C_6H_6$  and petrol-CHCl<sub>3</sub> with gradually increasing polarity) gave in order of elution, 2a (16 mg), 1a (35 mg), 1b (143 mg) and sitosterol (31 mg). Fractions eluted with CHCl<sub>3</sub> were separated and purified by successive prep. TLC on silica gel HF<sub>254</sub>. The developing solvents were different mixtures of petrol, hexane, EtOAc and Me<sub>2</sub>CO and gave in order of elution, 5a and 6a-component mixture (79 mg), 7a (23 mg), 3a (66 mg), 4a (17 mg), 7b (18 mg) and 8a (43 mg).

Identification of known compounds. The known compounds were identified by direct comparison of mp,  $[\alpha]_D$  and spectral data with published data (1b, 2a, 3a, 4a, 5a, 6a and 7a). For these compounds only new data are included.

Rel-(7S,8S)- $\Delta^{*}$ -6'-acetoxy-5'-methoxy-3,4-methylenedioxy-7-O-2',8.3'-neolignan (1c). (Ac<sub>2</sub>O-pyridine, 24 hr room temp.). Oil. IR  $v_{\text{max}}^{\text{max}}$  cm<sup>-1</sup>: 3082, 2963, 2920, 2860, 1763, 1638, 1612, 1502, 1440, 1368, 1346, 1250, 1190, 1115, 1030, 990, 930, 889, 854, 755. 

<sup>1</sup>H NMR (60 MHz, CCl<sub>4</sub>);  $\delta$ 6.60-6.88 (3H, m, H-2, H-5 and H-6), 4.93 (1H, d, J = 8.0 Hz, H-7), 3.07-3.90 (1H, m, H-8), 1.36 (3H, d, J = 7.0 Hz, H-9), 6.46 (1H, s, H-4'), 3.20 (2H, d, J = 6.0 Hz, H-7'), 5.66-6.20 (1H, m, H-8'), 4.66-5.30 (2H, m, H-9'), 5.88 (2H, s, OCH<sub>2</sub>O), 3.73 (3H, s, OMe-5'), 2.20 (3H, s, OAc-6'). MS 70 eV, m/z (rel. int.): 382 [M]\* (36), 341 (14), 340 (100), 325 (18), 311 (2), 297 (9), 249 (3), 179 (4), 162 (2), 135 (15).

Rel-(7S,8S)- $\Delta^{8}$ -6'-methoxy-3,4-methylenedioxy-7-O-3',8.4',1'-O-7'-neolignan (1b). [a] $_{20}^{20^{\circ}C}$  = +6.90° (CHCl<sub>3</sub>; c 4.2). <sup>13</sup>C NMR (20 MHz, CDCl<sub>3</sub>):  $\delta$ 144.28 (s, C-1), 106.37 (d, C-2), 147.82 (s, C-3), 147.33 (s, C-4), 107.84 (d, C-5), 119.44 (d, C-6), 92.59 (d, C-7), 45.64 (d, C-8), 18.12 (q, C-9), 134.76 (s, C-1), 97.28 (d, C-2'), 148.88 (s, C-3'), 122.25 (s, C-4'), 109.35 (d, C-5'), 153.31 (s, C-6'), 69.94 (t, C-7'), 133.41 (d, C-8'), 117.29 (t, C-9'), 100.76 (t, OCH<sub>2</sub>O), 57.19 (q, OMe-6').

Rel-(7R,8R,1'S,3'S)- $\Delta^{5+8}$  -4-hydroxy-3,5'-dimethoxy-1',2',3',4'-tetrahydro-2',4'-dioxo-7.3',8.1'-neolignan (7a). Oil. [ $\alpha$ ] $_D^{5*C}$  =  $-0.60^{\circ}$  (CHCl $_3$ ; c 0.83). Acetate of 7a (Ac $_2$ O-pyridine, 24 hr, room temp.). Oil. IR  $\nu_{\rm max}^{\rm him}$  cm  $^{-1}$ : 3060, 2920, 2860, 1760, 1713, 1620, 1530, 1515, 1455, 1436, 1375, 1275, 1200, 1160, 1120, 1040, 1003, 920, 860.  $^{1}$ H NMR (60 MHz, CCl $_4$ ):  $\delta$ 6.50-6.90 (3H, m, H-2, H-5 and H-6), 3.81 (3H, s, OMe-3), 2.23 (3H, s, OAc-4), 2.68 (1H, d, d) = 8.0 Hz, H-7), 1.83-2.45 (1H, d), H-8), 0.98 (3H, d) = 7.0 Hz, H-9), 3.42 (1H, d), H-3'), 3.62 (3H, d), OMe-5'), 5.66 (1H, d), 2.45-2.87 (2H, d), H-7'), 5.65-6.05 (1H, d), H-8'), 4.93-5.55 (2H, d), H-9'). MS 70 eV, d(d) (10), 178 (7), 137 (12).

Rel-(7R,8R,1'S,3'S)- $\Delta^{5',8'}$ -5'-methoxy-3,4-methylenedioxy-1',2',3',4'-tetrahydro-2',4'-dioxo-7.3',8.1'-neolignan (7b). Oil. [ $\alpha$ ] $_{\rm B}^{\rm SC}$ C = -7.31° (CHCl $_{\rm 3}$ ; c 0.20). UV  $\lambda$   $_{\rm max}^{\rm MeOH}$  nm (log  $\varepsilon$ ): 236

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(3.60), 289 (3.46). IR  $v_{max}^{\text{dim}}$  cm<sup>-1</sup>: 3085, 2985, 2940, 2860, 1770, 1718, 1650, 1618, 1515, 1498, 1452, 1390, 1335, 1265, 1245, 1225, 1187, 1114, 1088, 1045, 1005, 938, 895, 872, 760. <sup>1</sup>H NMR (60 MHz, CCl<sub>4</sub>):  $\delta$ 6.42–6.79 (3H, m, H-2, H-5 and H-6), 2.47 (1H, d, J = 7.3 Hz, H-7), 1.77–2.20 (1H, m, H-8), 1.04 (3H, d, J = 6.6 Hz, H-9), 3.56 (1H, s, H-3'), 5.64 (1H, s, H-6'), 2.25–3.05 (2H, m, H-7'), 5.60–6.12 (1H, m, H-8'), 4.95–5.36 (2H, m, H-9'), 3.67 (3H, s, OMe-5'), 5.92 (2H, s, OCH<sub>2</sub>O). MS 70 eV, m/z (rel. int.): 340 [M]\* (100), 325 (6), 312 (2), 299 (27), 284 (10), 271 (27), 243 (5), 178 (5), 162 (47), 150 (4), 137 (6), 135 (25).

Rel-(7S,8S,1'S,2'S,3'R,4'S)- $\Delta^{11}$ -2',4'-dihydroxy-3,4-methylenedioxy-1',2',3',4',5',6'-hexahydro-5'-oxo-7.3',8.1'-neolignan (8a). Crystals, mp 158–160° (CCl<sub>4</sub>). [ $\alpha$ ] $_{D}^{25^{\circ}C}$  = -0.69° (Me<sub>2</sub>CO; c 2.15). UV  $\lambda$  $_{E10}^{E10H}$  nm (log s): 237 (3.58), 289 (3.51). IR  $\nu$  $_{E10}^{KBF}$  cm  $^{-1}$ : 3504, 3450, 3378, 3078, 3014, 2956, 2915, 1698, 1640, 1610, 1501, 1485, 1451, 1390, 1321, 1250, 1210, 1185, 1152, 1118, 1100, 1070, 1040, 1015, 985, 963, 938, 921, 892, 880, 795, 730.  $^{1}$ H NMR (60 MHz, Me<sub>2</sub>CO-d<sub>6</sub>): see Table 1;  $^{13}$ C NMR (20 MHz, Me<sub>2</sub>CO-d<sub>6</sub>) and APT experiment, see Table 1. MS 30 eV, m/z (rel. int.): 330 [M] $^+$  (100), 271 (6), 240 (15), 199 (3), 177 (7), 149 (3), 162 (13), 135 (4).

 $Rel-(7S,8S,1'S,2'S,3'R,4'S)-\Delta^{5',8'}-2',4',5'-triacetoxy-3,4-methyl$ enedioxy-1',2',3',4'-tetrahydro-7.3',8.1'-neolignan (8b). (Ac<sub>2</sub>Opyridine, 120 hr room temp.). Oil. IR  $v_{\text{max}}^{\text{film}}$  cm<sup>-1</sup>: 3074, 3019, 2960, 2920, 2872, 1748 vs, 1668, 1641, 1610, 1502, 1486, 1442, 1368, 1220 s, 1035, 978, 923, 858, 810, 790, 748. <sup>1</sup>H NMR (60 MHz, CDCl<sub>3</sub>): δ6.70-6.95 (3H, m, H-2, H-5 and H-6), 2.10-2.80 (2H, m, H-7 and H-8), 0.99 (3H, d, J = 7.0 Hz, H-9), 5.60 (1H, br s, H-2'), 2.10-2.80 (1H, m, H-3'), 4.88-5.08 (1H, m, H-4'), 5.30 (1H, br s, H-6'), 1.70-2.10 (2H, m, H-7'), 5.55-6.10 (1H, m, H-8'), 4.88-5.55 (2H, m, H-9), 5.94 (2H, s, OCH<sub>2</sub>O), 2.08 (3H, s, OAc-2'), 2.13 (3H, s, OAc-4'), 2.16 (3H, s, OAc-5'). 13C NMR (20 MHz, CDCl<sub>3</sub>):  $\delta$ 134.20 (s, C-1), 108.10 (d, C-2), 146.10 (s, C-3), 143.84 (s, C-4), 108.20 (d, C-5), 125.97 (d, C-6), 53.02 (d, C-7), 50.83 (d, C-8), 12.75 (q, C-9), 50.65 (s, C-1'), 73.77 (d, C-2'), 53.02 (d, C-3'), 77.60 (s, C-4'), 147.93 (s, C-5'), 121.44 (d, C-6'), 36.10 (t, C-7'), 137.08 (d, C-8'), 117.76 (t, C-9'), 100.91 (t, OCH<sub>2</sub>O), 20.86 (q, MeCO-2'), 21.04 (q, MeCO-4'), 21.29 (q, MeCO-5'), 169.38 (s, MeCO-2'), 169.79 (s, MeCO-4'), 170.39 (s, MeCO-5'). MS 30 eV, m/z (rel.

int.): 456 [M]\* (100), 414 (5), 355 (5), 354 (9), 336 (9), 296 (3), 295 (11), 294 (31), 276 (8), 253 (9), 162 (59), 135 (7).

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#### REFERENCES

- Bernardi, L. (1962) Lauraceas, pp. 26 and 85. Talleres gráficos Universitarios, Mérida, Venezuela.
- Silva, R., Nagem, T. J., Mesquita, A. A. L. and Gottlieb, O. R. (1983) Phytochemistry 22, 772.
- Acero, L. E. (1979) Principales plantas útiles de la Amazonia Colombiana. Proyecto Radargramétrico del Amazonas: Editora Guadalupe Ltda., Bogotá, Colombia.
- 4. Gottlieb, O. R. (1978) Prog. Chem. Org. Nat. Prod. 35, 1.
- Gottlieb, O. R., Da Silva, M. L. and Ferreira, Z. S. (1975) Phytochemistry 14, 1825.
- Gottlieb, O. R., Mourão, J. C., Yoshida, M., Mascarenhãs, Y. P., Rodrigues, M., Rosenstein, R. D. and Tomita, K. (1977) Phytochemistry 16, 1003.
- Araujo Lima, O., Gottlieb, O. R. and Magalhães, M. T. (1972) Phytochemistry 11, 2031.
- Alvarenga, M. A. de, Brockson, U., Castro, C. O., Gottlieb,
   O. R. and Magalhães, M. T. (1977) Phytochemistry 16, 1797.
- Martínez, V. J. C., Maia, J. G. S., Yoshida, M. and Gottlieb, O. R. (1980) Phytochemistry 19, 474.
- Alvarenga, M. A. de, Castro, O. C., Giesbrecht, A. M. and Gottlieb, O. R. (1977) Phytochemistry 16, 1801.
- Haraguchi, M., Motidome, M., Yoshida, M. and Gottlieb, O. R. (1983) Phytochemistry 22, 561.
- Gomes, M. C. C. P., Yoshida, M., Gottlieb, O. R., Martínez, J. C. and Gottlieb, H. E. (1983) Phytochemistry 22, 269.
- Giesbrecht, A. M., Franca, N. C. and Gottlieb, O. R. (1974) Phytochemistry 13, 2285.